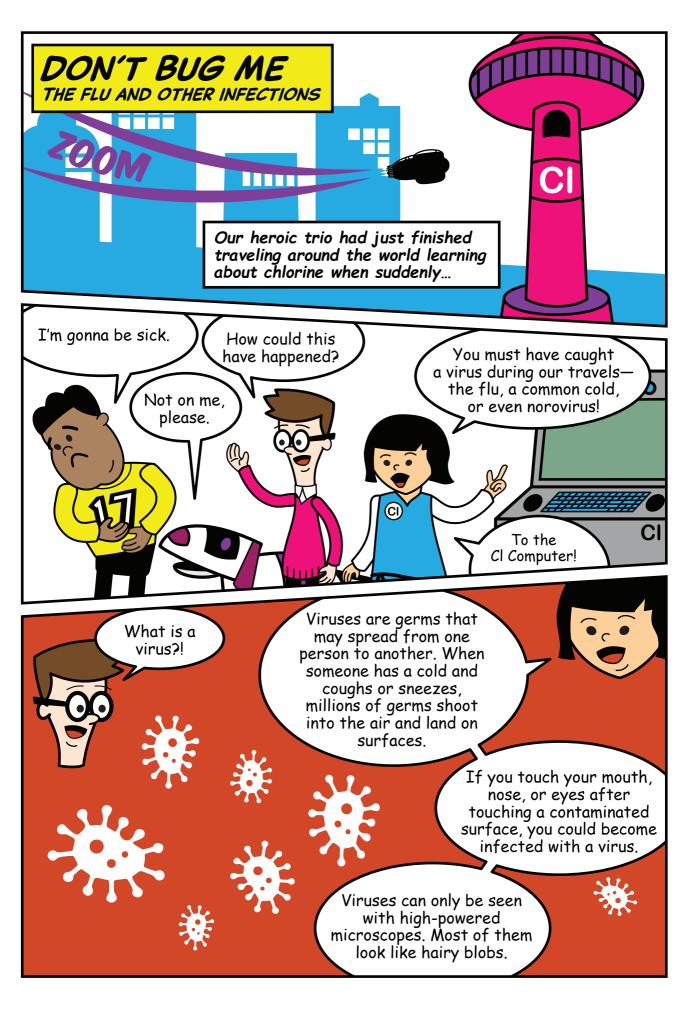
THE ADVENTURES OF VOLUME #2: DISINFECTION CALLS TE AND INTRODUCING: CLETUS, THE ROBO-DOG

THE ADVENTURES OF TEAM CHLORINE

Team Chlorine is a trio of students who study the element chlorine. Why on earth are they so interested in chlorine, you may be wondering?





Everyone chill! Oh, no! I hope Here are things this virus we can all do to doesn't keep spreading! help prevent that from happening: 1. Wash your hands often with warm water and soap. Remember to scrub between your fingers and wash for as long as it takes you to sing the "Happy Birthday 20 Song" twice. sec **2.** Cover up coughs and sneezes with a tissue or the inside of your elbow. Be sure to throw the tissue away immediately and wash your hands! Don't touch your eyes, nose, or mouth. **3.** Don't touch your eyes, hose, or an your Germs are good at hitching a ride on your hands to these spots where they can get into your body! 4. Adults can disinfect surfaces and help destroy germs by using a wipe or spray that contains bleach to kill germs on desks, door knobs, handrails, and other hard surfaces that a lot of people touch. Remember. I'm looking at washing your you, Clark! hands includes using soap. Great, Good. Keep resting, A few days later... now take though. It's important me for a to stay home until walk you're healthy before I'm starting going back to school to feel much or work. better.



Find more activities and answers to the ones in this book at teamchlorine.com.

Match the man with the chlorine history he made!

1. Carl-Wilhelm Scheele

2. Humphry Davy

3. John Leal A. Advocated for water chlorination

B. Proved chlorine is an element

C. Discovered chlorine

Match the compound!

One reason why chlorine is such a valuable element is because it bonds very easily with other elements to form compounds. Match each compound below with its symbol, using the Periodic Table of Elements on the back cover to help you.

- Sodium chloride Helps make your food taste great
 Hydrogen chloride Your body makes it to help digest food
 - **3. Titanium dioxide** Used to make paint pigments
 - 4. Calcium chloride Makes icy roads less slippery
 - 5. Sodium hypochlorite Also known as household bleach
 - 6. Ferric chloride Helps keep drinking water clean

NaOCI

Α.

B. FeCl₃

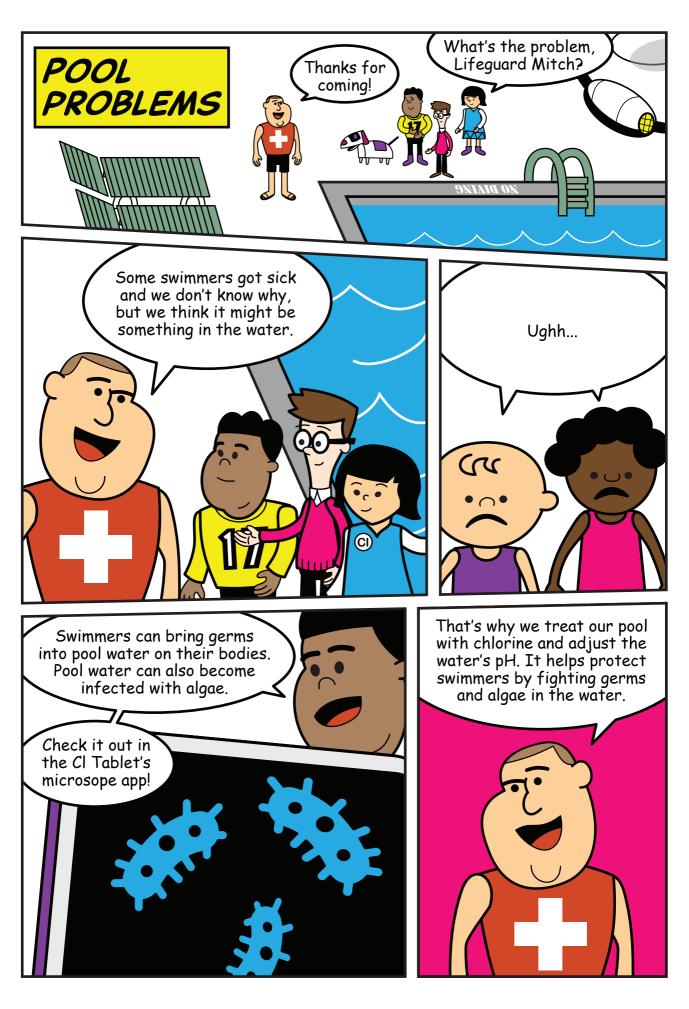
C. NaCl

D. HCI

E. TiO₂

F. CaCl₂

Break the code! Using the Periodic Table of Elements on the back cover, match the numbers below to the corresponding element symbol to discover the answer. Question: What makes chlorine so amazing?
Answer: Chlorine 20 7 49 16 15 53 75 66 7 95 53 6 49 7 8 23 85 53 8 7
Mixed-up tale! Fill in the blanks to create a silly story about healthy pool water. It was a beautiful summer day to go for a I [noun] . I [verb (past tense)] on my swim suit and headed for the town Before I got into the
cool, refreshing, I made sure to go and take a so I wouldn't [noun] [noun] [noun] [noun] [noun] [name of girl] was supposed to meet me there, but it turned out she had a[adjective]
stomach ache so she stayed home. It's not a good to go swimming if you are because you canyour germs into the water [adjective] and make other sick. I in the pool for a long time and the only reason I got out was to go and use the It's very important that all swimmers keep pool water [adjective] by using the bathroom.



Of course! But pH and chlorine levels need to be monitored. Always check that pool water is healthy for swimming by using your senses:

Look at the water.

Does the pool water look clear and blue? You should be able to see through the water down to the drain or stripes painted on the floor of the pool. If the water is cloudy and colored, there may be algae in it. Don't go in!

Touch the side of the pool.

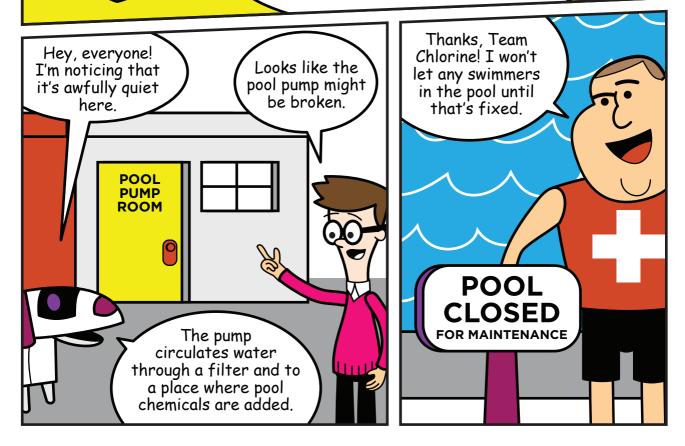
2. Does the pool wall around the water line feel slimy? If it does, there are probably germs living on the wall. Don't go in!

3. Smell the pool area. Is there a strong chemical odor around the pool? If there is, the pool manager may have to treat the water. Don't go in!

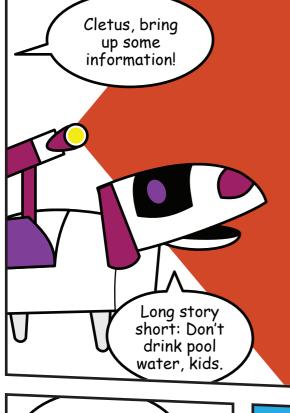
4. Listen for noise.

The sound of a pool pump is a good sign!

Of course, don't ever use your sense of taste when it comes to pools. If you do get some water in your mouth, don't swallow it.









Common viruses that infect pool water

Pseudomonas aeruginosa

[soo-do-mo'-nuhs air-ooh-gi-no'-sa]

- Fast-moving bacteria with a whip-like tail
- Enters pool on swimmers' bodies or in dirt tracked into the pool area
- Causes a bumpy, itchy skin rash and an earache known as "swimmer's ear"

Shigella sonnei

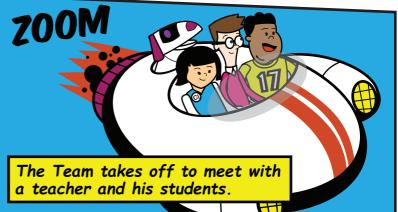
[shi-gell'-uh so'-nay]

- Rod-shaped bacteria
- Enters pool water by swimmers who use the bathroom and don't wash their hands
- Causes diarrhea if swallowed

Cyptosporidium parvum

[sip-to-spor-ih'-dee-um par'-vum]

- Microscopic parasite surrounded by a protective shell
- Enters pool water in the poop of infected swimmers
- Causes diarrhea if swallowed



Did you know?

When your eyes turn red after being in a pool, it's not from the chlorine. In fact, it may indicate that the pool needs more chlorine.

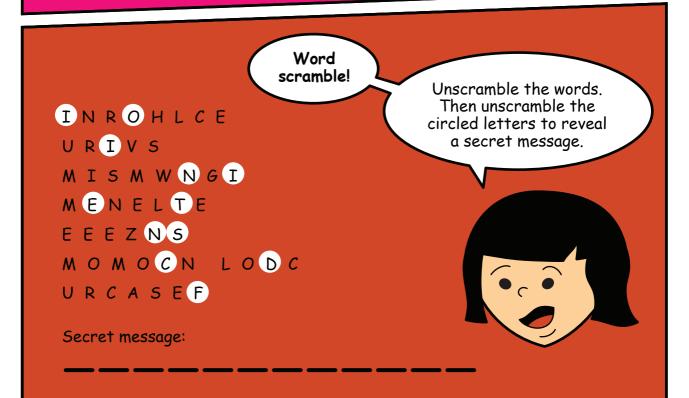
Happy to report that chlorine doesn't turn hair green, either!

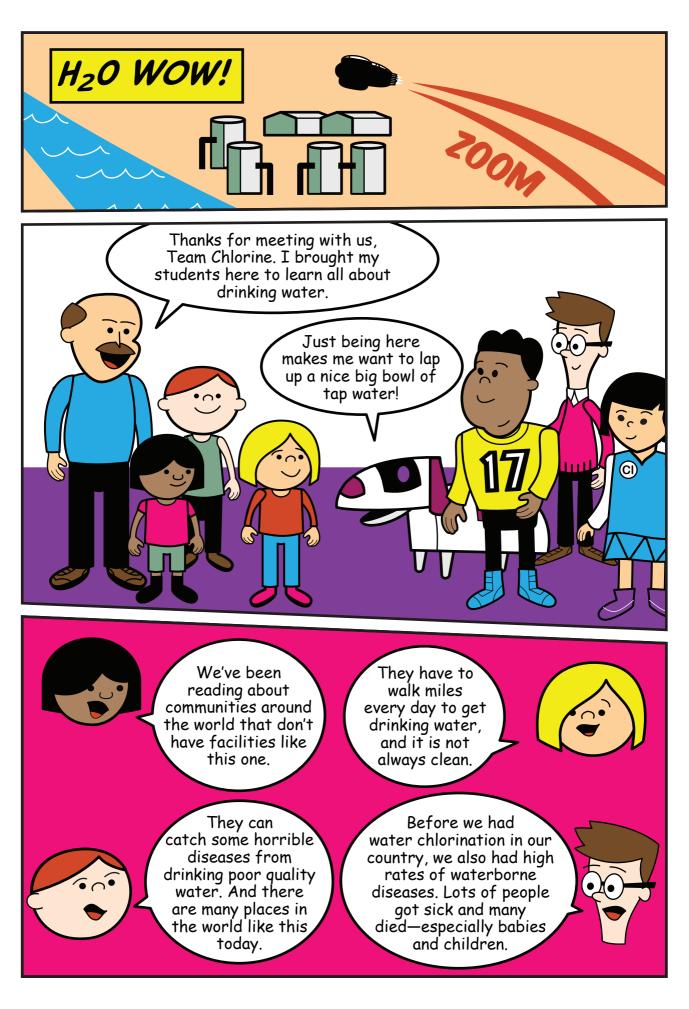
17 ^{35.453} **C** Chlorin<u>e</u>

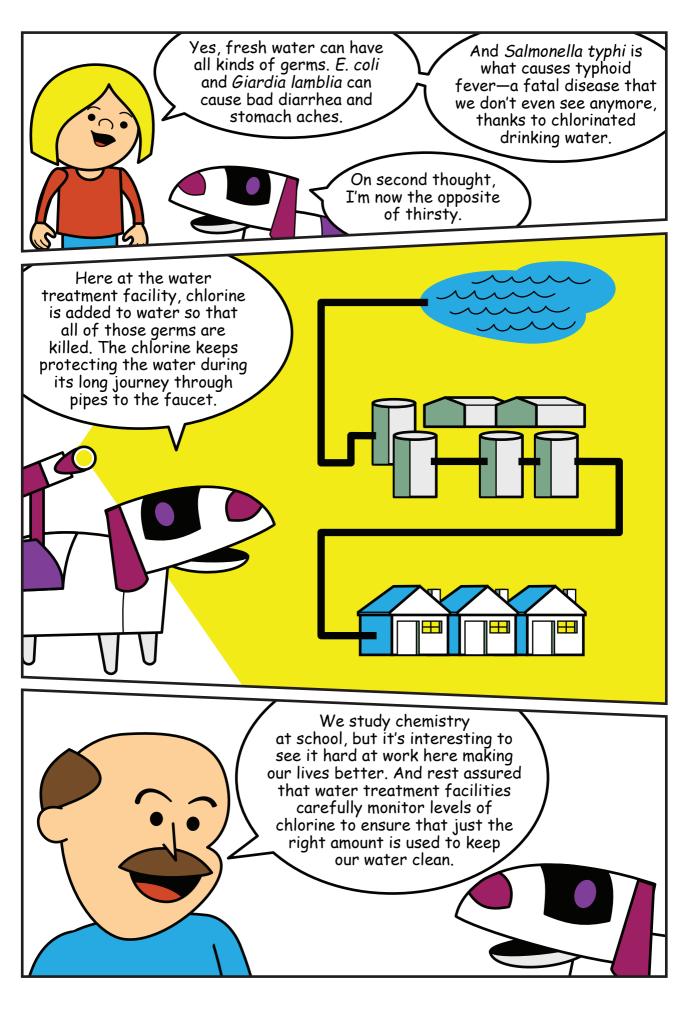
A quick history lesson

Chlorine was discovered in 1774 by the Swedish pharmacist and chemist Carl-Wilhelm Scheele while he was studying the mineral pyrolusite. Scheele mixed the mineral (which is the chemical compound manganese dioxide) with hydrochloric acid and produced chlorine gas. He noticed that the gas reacted with metals. however his discovery was not understood and it was believed the gas was a compound of oxygen.

Chlorine was not proven to be an element until 1810 by English chemist Humphry Davy.







DID YOU KNOW?

In 2010 the United Nations declared that clean drinking water and water sanitation are at the core of all human rights. Without access to these essentials, poverty and disease thrive.



In 2015, the United Nations announced Sustainable Development Goals—17 ways to ensure a sustainable future for people everywhere. Providing safe drinking water and wastewater sanitation is one of those goals.

Read up on how chlorinated drinking water has been saving lives for over 100 years.

The history of drinking water chlorination

At the beginning of the 20th century in America, waterborne diseases like typhoid fever and dysentery were rampant. Deaths from these diseases were very common.

Cities and industries were thriving, and newly built sewer systems were infecting drinking water sources. A pioneering physician named Dr. John L. Leal thought water chlorination was the answer. He knew chlorine was a safe and effective disinfectant used to help control some infectious disease outbreaks.

Thanks to Dr. Leal's persistence, residents of Jersey City, New Jersey became the first Americans to have continuous access to chlorinated municipal drinking water, starting in 1908. Within a decade, 33 million people across the country did. Rates of waterborne disease plummeted and life expectancy increased dramatically.









Find the words below that are related to chlorine. Some are products that are made with chlorine chemistry.

pharmaceutical sunscreen element paint disinfectant chlorine battery water solar panel compound bleach pool

Word

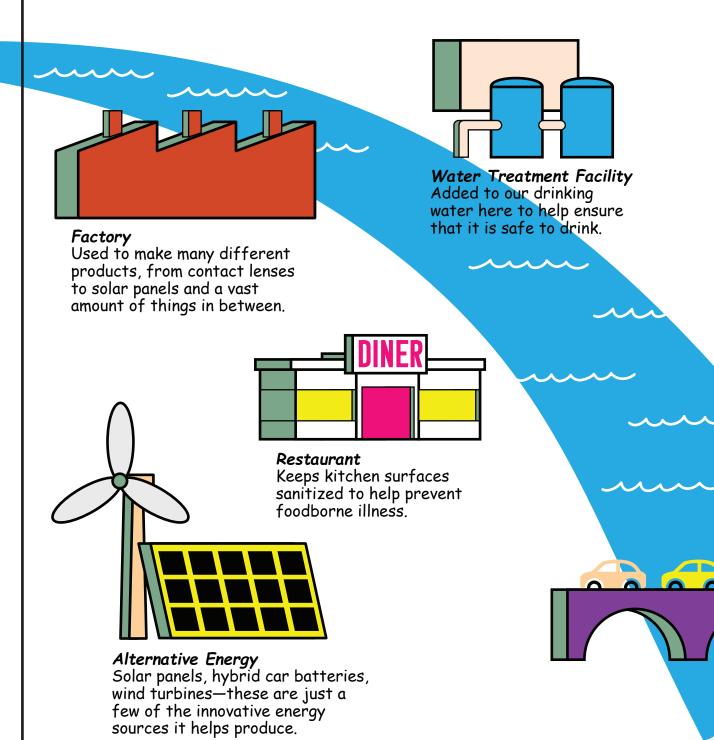
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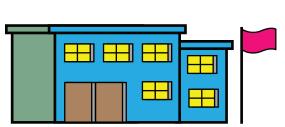
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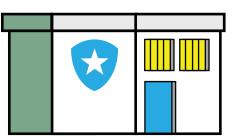
CHLORINE IS PART OF YOUR COMMUNITY

The element chlorine helps make life better in many different ways. That's because chlorine bonds easily with many other elements, creating all kinds of "chlorine chemistry" that is hard at work every day in a lot of places you go.





School Keeps surfaces sanitized to help prevent the spread of flu and other diseases.

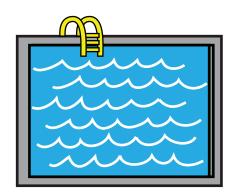


Police Station Helps make many of the important products police officers rely on, including bullet-resistant vests, helmets, and glass.

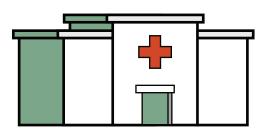


Home

Brings safe drinking water to your faucet and is used to help make all kinds of products in your home—windows, gutters, siding, flooring, paint, and more.



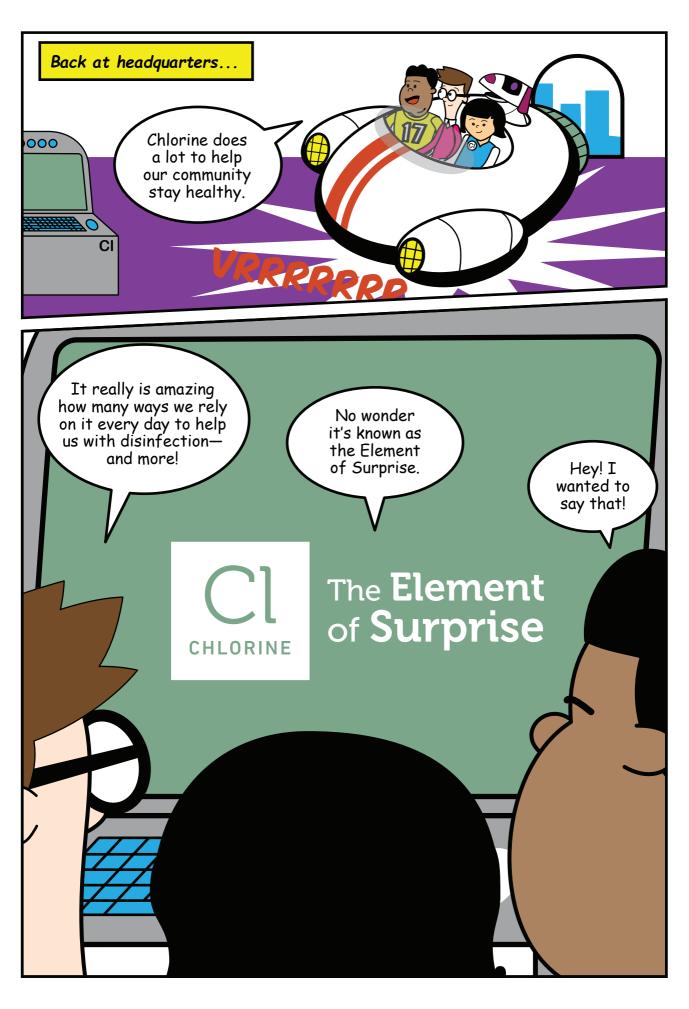
Pool Ensures that pool water is sanitized and healthy for swimming.



Hospital

Essential to disinfection and key in producing healthcare products like pharmaceuticals, inhalers, blood bags, tubings, medical devices, and more.

Transportation Cars, trains, planes, ships, and even spacecraft all rely on it for many of their materials and parts.







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